

SHORT  
COMMUNICATIONS

## Recyclization of Pyrroloquinoxalinetriene by the Action of *o*-Aminobenzenethiol

K. S. Bozdyreva and A. N. Maslivets

Perm State University, ul. Bukireva 15, Perm, 614990 Russia  
e-mail: koh2@psu.ru

Received September 21, 2005

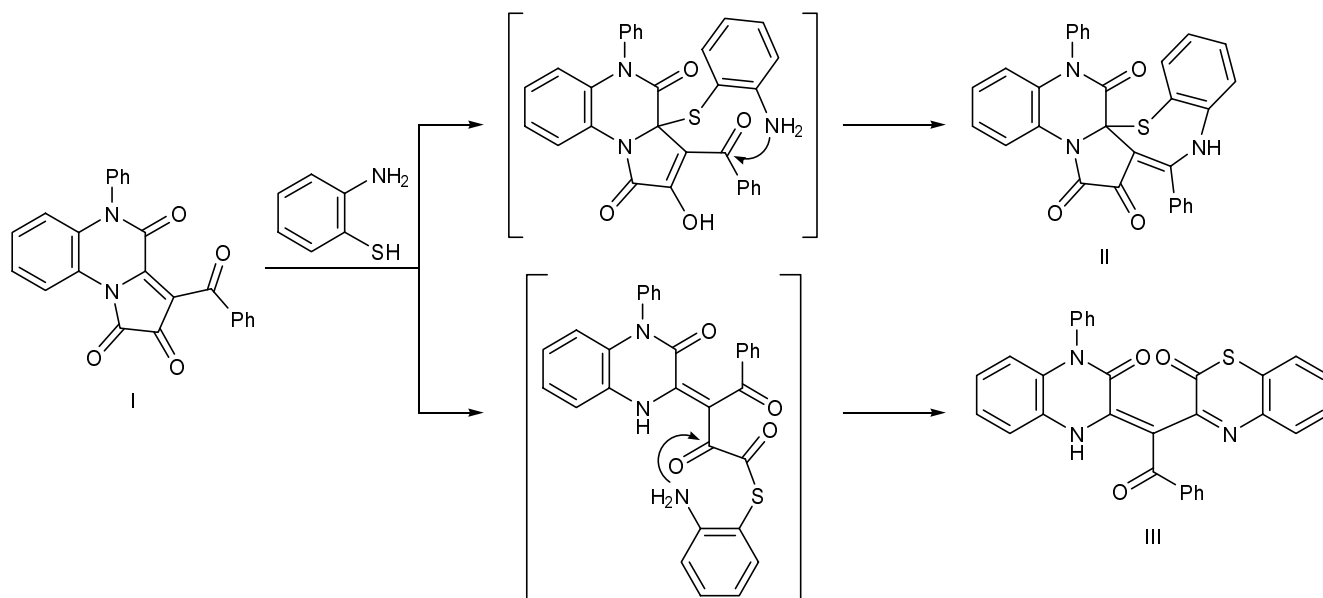
DOI: 10.1134/S1070428006030249

We previously showed that the reaction of 3-aryl-5-phenylpyrrolo[1,2-*a*]quinoxaline-1,2,4(5*H*)-triones with *o*-aminobenzenethiol involves successive attack by the sulfanyl and amino groups of the latter on the carbon atom in position 3*a* and aryl carbonyl carbon atom, respectively, of the former to give heterocyclization products, 8-aryl-16-phenyl-6*H*-quinoxalino[1',2':1,2]pyrrolo[2,3-*b*][1,5]benzothiazepine-6,7,15(9*H*,16*H*)-triones [1].

In the reaction of 3-benzoyl-5-phenylpyrrolo[1,2-*a*]quinoxaline-1,2,4(5*H*)-trione (**I**) with *o*-aminobenzenethiol in boiling benzene (reaction time 10 min), we isolated the expected product, 8,16-diphenyl-6*H*-quinoxalino[1',2':1,2]pyrrolo[2,3-*b*][1,5]benzothiazepine-6,7,15(9*H*,16*H*)-trione (**II**), and 3-[(1*Z*)-2-oxo-1-(3-oxo-4-phenyl-3,4-dihydroquinoxalin-2(1*H*)-ylidene)-2-phenylethyl]-1,4-benzothiazin-2(3*H*)-one (**III**).

Presumably, benzothiazinone **III** is formed as a result of attack by the sulfanyl group of *o*-aminobenzenethiol on the C<sup>1</sup> carbon atom of **I** with opening of the pyrroledione ring at the C<sup>1</sup>-N<sup>10</sup> bond and subsequent intramolecular cyclization involving the amino group of aminobenzenethiol and carbonyl group in the β-position with respect to the sulfur atom. Among several possible tautomeric forms of benzothiazinone **III**, that giving rise to the strongest intramolecular hydrogen bond is observed.

**8,16-Diphenyl-6*H*-quinoxalino[1',2':1,2]pyrrolo[2,3-*b*][1,5]benzothiazepine-6,7,15(9*H*,16*H*)-trione (**II**) and 3-[(1*Z*)-2-oxo-1-(3-oxo-4-phenyl-3,4-dihydroquinoxalin-2(1*H*)-ylidene)-2-phenylethyl]-1,4-benzothiazin-2(3*H*)-one (**III**).** A solution of 1 mmol of compound **I** and 1 mmol of *o*-aminobenzenethiol in 10 ml of anhydrous benzene was heated for 10 min



under reflux. The mixture was cooled, and the light yellow precipitate of compound **III** was filtered off. Yield 25%, mp 284–286°C (decomp., from benzene). IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3030 br (NH), 1728 (SC=O), 1703 (NC=O), 1630 (PhCO).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 6.57 d (1H, 5-H, quinoxaline,  $J = 7.5$  Hz), 7.15–7.98 m (17H;  $\text{H}_{\text{arom}}$ ; 6-H, 7-H, and 8-H in quinoxaline, 5-H, 6-H, 7-H, and 8-H in benzothiazine), 13.77 s (1H, NH).  $^{13}\text{C}$  NMR spectrum (DMSO- $d_6$ ),  $\delta_{\text{C}}$ , ppm: 115.97–140.04 ( $\text{C}_{\text{arom}}$ ), 147.17 (NC=O), 156.15 ( $\text{C}^3$ , benzothiazine), 164.39 (SC=O), 188.68 (PhCO). Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 501 (73)  $[\text{M}]^+$ , 445 (9)  $[\text{M} - 2\text{CO}]^+$ , 424 (100)  $[\text{M} - \text{Ph}]^+$ , 396 (88)  $[\text{M} - \text{Ph} - \text{CO}]^+$ , 368 (35)  $[\text{M} - \text{Ph} - 2\text{CO}]^+$ , 323 (21), 262 (87), 234 (27), 212 (34), 205 (38), 109 (25)  $[\text{PhS}]^+$ , 77 (28)  $[\text{Ph}]^+$ . Found, %: C 71.86; H 3.85; N 8.36; S 6.37.  $\text{C}_{30}\text{H}_{19}\text{N}_3\text{O}_3\text{S}$ . Calculated, %: C 71.84; H 3.82; N 8.38; S 6.39.

The mother liquor was evaporated to dryness. The residue was dark yellow compound **II**. Yield 64%, mp 239–241°C (decomp., from toluene). IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3210 br (NH), 1732 ( $\text{C}^6=\text{O}$ ), 1685 ( $\text{C}^7=\text{O}$ ), 1670 ( $\text{C}^{15}=\text{O}$ ).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 6.41 d (1H, 1-H,  $J = 7.5$  Hz), 7.16–7.56 m (16H,  $\text{H}_{\text{arom}}$ , 2-H, 3-H, 10-H, 11-H, 12-H, 13-H), 7.91 d (1H, 4-H,  $J = 7.7$  Hz),

10.40 s (1H, NH).  $^{13}\text{C}$  NMR spectrum (DMSO- $d_6$ ),  $\delta$ , ppm: 67.98 ( $\text{C}^{14\text{a}}$ ), 105.82 ( $\text{C}^{7\text{a}}$ ), 116.40–137.32 ( $\text{C}_{\text{arom}}$ ), 143.64 ( $\text{C}^6$ ), 156.60 ( $\text{C}^{15}$ ), 159.49 ( $\text{C}^8$ ), 174.23 ( $\text{C}^7$ ). Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 501 (23)  $[\text{M}]^+$ , 445 (55)  $[\text{M} - 2\text{CO}]^+$ , 424 (11)  $[\text{M} - \text{Ph}]^+$ , 396 (21)  $[\text{M} - \text{Ph} - \text{CO}]^+$ , 368 (6)  $[\text{M} - \text{Ph} - 2\text{CO}]^+$ , 262 (100), 234 (23), 223 (11), 212 (46), 109 (28), 77 (20)  $[\text{Ph}]^+$ . Found, %: C 71.88; H 3.80; N 8.34; S 6.40.  $\text{C}_{30}\text{H}_{19}\text{N}_3\text{O}_3\text{S}$ . Calculated, %: C 71.84; H 3.82; N 8.38; S 6.39.

The IR spectra were recorded on a UR-20 spectrometer from samples dispersed in mineral oil. The  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra were measured on a Bruker WP-400 instrument from solutions in DMSO- $d_6$  using tetramethylsilane as internal reference. The purity of the products was checked by thin-layer chromatography on Silufol plates using ethyl acetate as eluent; spots were visualized by treatment with iodine vapor.

This study was performed under financial support by the Russian Foundation for Basic Research (project no. 04-03-96033).

## REFERENCE

1. Maslivets, A.N. and Bozdyreva, K.S., *Khim. Geterotsikl. Soedin.*, 2002, no. 12, p. 1735.